

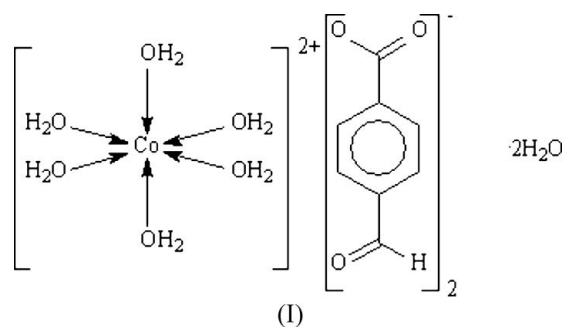
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Key indicators

Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
Disorder in main residue
 R factor = 0.029
 wR factor = 0.080
Data-to-parameter ratio = 13.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Hexaaquacobalt(II) bis(4-formylbenzoate)
dihydrateThe crystal structure of the title compound, $[\text{Co}(\text{H}_2\text{O})_6](\text{C}_8\text{H}_5\text{O}_3)_2 \cdot 2\text{H}_2\text{O}$, consists of octahedral Co^{II} complex cations that interact with the anions and uncoordinated water molecules, forming a tightly held, hydrogen-bonded layer. The Co^{II} complex cation lies on a special position of site symmetry $\bar{1}$.Received 6 November 2006
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Comment

The known transition metal derivatives of 4-formylbenzoic acid are water-coordinated compounds showing covalent metal–carboxylate bonds, for example the nickel(II) (Deng *et al.*, 2006a) and copper(II) (Deng *et al.*, 2006b) derivatives. However, the cobalt(II) derivative does not have this feature; the title compound (I) has a hexaaquacobalt cation (Table 1), and its charge is balanced by two uncoordinated 4-formylbenzoate ions. There are one cation and two anions in the asymmetric unit along with two uncoordinated water molecules (Fig. 1). These are linked by hydrogen bonds (Table 2) into a layer structure. The Co^{II} complex cation lies on a special position of site symmetry $\bar{1}$.

Experimental

Cobalt diacetate trihydrate (0.116 g, 0.5 mmol) was added to an aqueous solution (20 ml) of 4-formylbenzoic acid (0.15 g, 1 mmol). Sodium hydroxide solution was added to obtain a pH 5. The filtered solution was allowed to stand for a few days for the pink prismatic crystals to separate. Elemental analysis calculated for $\text{C}_{16}\text{H}_{26}\text{CoO}_{14}$: C 38.33, H 5.23%; found: C 38.31, H 5.24%.

Crystal data

 $[\text{Co}(\text{H}_2\text{O})_6](\text{C}_8\text{H}_5\text{O}_3)_2 \cdot 2\text{H}_2\text{O}$
 $M_r = 501.30$
 Triclinic, $P\bar{1}$
 $a = 6.3470$ (3) Å
 $b = 7.4194$ (4) Å
 $c = 12.5041$ (7) Å
 $\alpha = 92.353$ (2)°
 $\beta = 90.029$ (3)°
 $\gamma = 113.801$ (6)°

 $V = 538.20$ (5) Å³
 $Z = 1$
 $D_x = 1.547$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.87$ mm⁻¹
 $T = 295$ (2) K
 Prism, pink
 $0.36 \times 0.28 \times 0.18$ mm

Data collection

Rigaku RAXIS-RAPID IP diffractometer
 ω scans
 Absorption correction: multi-scan *ABSCOR* (Higashi, 1995)
 $T_{\min} = 0.614$, $T_{\max} = 0.860$

5333 measured reflections
 2451 independent reflections
 2325 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.017$
 $\theta_{\text{max}} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.029$
 $wR(F^2) = 0.080$
 $S = 1.00$
 2451 reflections
 187 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0485P)^2 + 0.2223P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.55 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\text{min}} = -0.27 \text{ e } \text{Å}^{-3}$

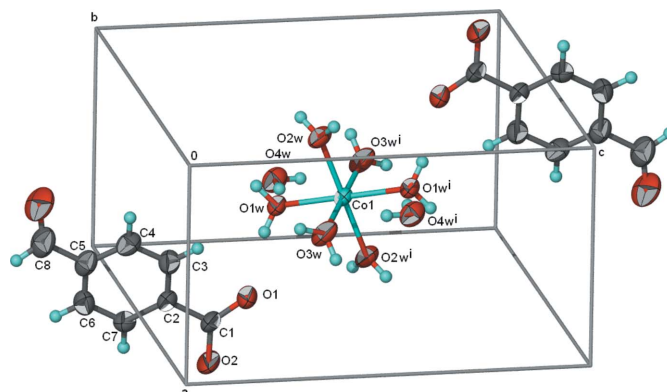


Figure 1
 Displacement ellipsoid plot of the unit-cell contents. Displacement ellipsoids are drawn at the 70% probability level, and H atoms are shown as spheres of arbitrary radii. Only one disorder component is shown. [Symmetry code: (i) $1 - x, 1 - y, 1 - z$.]

Table 1

Selected geometric parameters (Å, °).

Co1—O1w	2.127 (1)	Co1—O3w	2.065 (1)
Co1—O2w	2.071 (1)		
O1w—Co1—O2w	89.57 (5)	O2w—Co1—O3w	88.04 (6)
O1w—Co1—O3w	92.73 (6)		

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1w—H1w1...O1	0.84 (1)	1.95 (1)	2.778 (2)	166 (2)
O1w—H1w2...O2 ⁱ	0.85 (1)	1.78 (1)	2.629 (2)	178 (3)
O2w—H2w1...O1 ⁱ	0.85 (1)	1.90 (1)	2.746 (2)	171 (2)
O2w—H2w2...O4w ⁱⁱ	0.84 (1)	1.93 (1)	2.760 (2)	172 (2)
O3w—H3w1...O3 ⁱⁱⁱ	0.84 (1)	1.89 (1)	2.715 (2)	165 (3)
O3w—H3w2...O4w ^{iv}	0.84 (1)	1.92 (1)	2.743 (2)	169 (2)
O4w—H4w1...O1 ^v	0.84 (1)	2.03 (2)	2.812 (2)	155 (3)
O4w—H4w2...O1w	0.84 (1)	2.05 (1)	2.872 (2)	168 (2)

Symmetry codes: (i) $x - 1, y, z$; (ii) $x - 1, y - 1, z$; (iii) $-x + 1, -y + 1, -z$; (iv) $x, y - 1, z$; (v) $-x + 2, -y + 2, -z + 1$.

The $-C_6H_4-$ ring is disordered over two positions about the $C2 \cdots C5$ vector; the atoms C2 and C5 are ordered. The C—C distances in the ring were restrained to 1.39 (1) Å, and both components were restrained to be approximately planar. The displacement parameters of the primed atoms were set equal to those of the unprimed atoms. The occupancy factors refined to 0.52 (1):0.48 (1).

Carbon-bound H atoms were positioned geometrically (C—H 0.93 Å) and were included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The water H atoms were located in a difference Fourier map, and were refined with a distance restraint of O—H 0.85 (1) Å; their displacement parameters were freely refined.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2006).

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References

Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
 Deng, Z.-P., Gao, S. & Ng, S. W. (2006a). *Acta Cryst. E* **62**, m2904–m2905.
 Deng, Z.-P., Gao, S. & Ng, S. W. (2006b). *Acta Cryst. E* **62**, m2906–m2907.
 Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
 Rigaku (1998). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.
 Rigaku/MS (2002). *CrystalStructure*. Rigaku/MS Inc. The Woodlands, Texas, USA.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Westrip, S. P. (2006). *pubCIF*. In preparation.